metal-organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Gerd W. Rabe,^a* Mei Zhang-Preße^a and Glenn P. A. Yap^b

^aTechnische Universität München, Anorganisch-chemisches Institut, Lichtenbergstraße 4, 85747 Garching, Germany, and ^bUniversity of Ottawa, Department of Chemistry, 130 Louis Pasteur St., Ottawa, Ontario K1N 6N5, Canada

Correspondence e-mail: g.rabe@lrz.tu-muenchen.de

Key indicators

Single-crystal X-ray study T = 203 KMean $\sigma(C-C) = 0.010 \text{ Å}$ Disorder in main residue R factor = 0.068 wR factor = 0.120 Data-to-parameter ratio = 17.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Bis(μ -2,2"-dimethoxy-*m*-terphenyl)di- μ ₃-iodo-bis(tetrahydrofuran)tetralithium(I)

The molecular structure of a donor-functionalized terphenyllithium compound, $\{(\mu-2,2''-\text{dimethoxy-}m\text{-terphenyl})(\mu\text{-Li}) [\mu\text{-Li}(\text{tetrahydrofuran})I]_2$ or $[\text{Li}_4\text{I}_2(\text{C}_4\text{H}_8\text{O})_2(\text{C}_{20}\text{H}_{17}\text{O}_2)_2]$, is reported, featuring a dimeric and centrosymmetric arrangement in its solid-state structure. The molecular structure is comprised of a distorted *transoid* four-step ladder formed by four Li, two I, and two *ipso*-C atoms of the terphenyl moieties. The two methoxy functions of each terphenyl moiety coordinate to different Li atoms. One molecule of tetrahydrofuran coordinates to each of the Li atoms, which occupy terminal positions of the ladder-type arrangement.

Comment

In recent years, there have been a number of reports on crystal structure determinations of different terphenyl-lithium compounds. The solid-state structures of both hydrocarbonsoluble unsolvated compounds (Ruhlandt-Senge et al., 1993; Wehmschulte & Power, 1997), as well as of their Lewis-base adducts have been reported (Crittendon et al., 1999; Girolami et al., 1992; Olmstead & Power, 1991; Wegner et al., 2000). Such lithium derivatives are typically employed as transfer reagents. We have previously reported crystal structure determinations of tetrahydrofuran adducts of two terphenyllithium compounds (Rabe et al., 2000). In this context, we were interested in investigating the solid-state structure of a donorfunctionalized terphenyl-lithium compound and determined the structure of $bis(\mu-2,2''-dimethoxy-m-terphenyl)di-\mu_3$ iodo-bis(tetrahydrofuran)tetralithium(I), (I), which is reported here.



Experimental

Li(μ -2,2"-dimethoxy-*m*-terphenyl) was prepared by reaction of a suspension of 2 μ -2,2"-dimethoxy-*m*-terphenyl iodide in hexanes with equimolar amounts of a 1.6 *M* solution of *n*-butyllithium in hexanes (Rabe *et al.*, 2001). Iodine analysis of the resulting hexane- and toluene-washed Li(μ -2,2"-dimethoxy-*m*-terphenyl) revealed that the

 \odot 2002 International Union of Crystallography Printed in Great Britain – all rights reserved

Received 20 June 2002 Accepted 10 July 2002 Online 19 July 2002 lithium compound contained 1–2% iodine. Single crystals of (I) were obtained by slow evaporation of a solution of $\text{Li}(\mu-2,2''-\text{dimethoxy-}m\text{-terphenyl})$ with trace amounts of lithium iodide in toluene, in the presence of small amounts of tetrahydrofuran.

Crystal data

$$\begin{split} & \left[\text{Li}_{4}\text{I}_{2}(\text{C}_{4}\text{H}_{8}\text{O})_{2}(\text{C}_{20}\text{H}_{17}\text{O}_{2})_{2} \right] \\ & M_{r} = 1004.44 \\ & \text{Monoclinic, } P_{1}/c \\ & a = 10.1825 (14) \text{ Å} \\ & b = 14.863 (2) \text{ Å} \\ & c = 15.396 (2) \text{ Å} \\ & \beta = 92.009 (2)^{\circ} \\ & V = 2328.7 (5) \text{ Å}^{3} \\ & Z = 2 \end{split}$$

Data collection

Bruker (Siemens) 1K CCD area-
detector diffractometer
φ and ω scans
Absorption correction: multi-scan
(SADABS; Bruker, 2001)
$T_{\min} = 0.866, T_{\max} = 0.972$
14670 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.068$ $wR(F^2) = 0.120$ S = 1.034726 reflections 270 parameters $D_x = 1.433 \text{ Mg m}^{-3}$ Mo K\$\alpha\$ radiation Cell parameters from 21 reflections $\theta = 1.5-26.4^\circ$ $\mu = 1.40 \text{ mm}^{-1}$ T = 203 (2) KPlate, colourless $0.10 \times 0.10 \times 0.02 \text{ mm}$

4726 independent reflections 2429 reflections with $I > 2\sigma(I)$ $R_{int} = 0.128$ $\theta_{max} = 26.4^{\circ}$ $h = -12 \rightarrow 12$ $k = 0 \rightarrow 18$ $l = 0 \rightarrow 19$

 $\begin{array}{l} \mbox{H-atom parameters constrained} \\ w = 1/[\sigma^2(F_o{}^2) + (0.039P)^2] \\ \mbox{where } P = (F_o{}^2 + 2F_c{}^2)/3 \\ (\Delta/\sigma)_{\rm max} < 0.001 \\ \Delta\rho_{\rm max} = 0.58 \mbox{ e } {\rm \AA}{}^{-3} \\ \Delta\rho_{\rm min} = -0.66 \mbox{ e } {\rm \AA}{}^{-3} \end{array}$

The tetrahydrofuran atom C21 was disordered over two positions (C21/C21') with a refined site-occupancy distribution of 0.62:0.38, and was constrained with a common isotropic displacement parameter. The vicinal atom C22 was given the same split occupancy for proper assignment of idealized riding H atoms, with C22 and C22' constrained to a common position and anisotropic displacement parameter. Chemically equivalent bond distances were restrained to be similar between the disordered contributors.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 2001); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2001); software used to prepare material for publication: *SHELXTL*.



Figure 1

The molecular structure of the title compound, showing the atom-labeling scheme. Displacement ellipsoids are depicted at the 30% probability level. The minor-disorder contributors and H atoms have been omitted for clarity.

References

- Bruker (2001). SMART, SAINT, SADABS and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Crittendon, R. C., Beck, B. C., Su, J., Li, X.-W. & Robinson, G. H. (1999). Organometallics, 18, 156–160.
- Girolami, G. S., Riehl, M. E., Suslick, K. S. & Wilson, S. R. (1992). Organometallics, 11, 3907–3910.
- Olmstead, M. M. & Power, P. P. (1991). J. Organomet. Chem. 408, 1-6.
- Rabe, G. W., Bérubé, C. D. & Yap, G. P. A. (2001). *Inorg. Chem.* 40, 4780–4784.
- Rabe, G. W., Sommer, R. D. & Rheingold, A. L. (2000). Organometallics, **19**, 5537–5540.
- Ruhlandt-Senge, K., Ellison, J. J., Wehmschulte, R. J., Pauer, F. & Power, P. P. (1993). J. Am. Chem. Soc. 115, 11353–11357.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Wegner, G. L., Berger, R. J. F., Schier, A. & Schmidbaur, H. (2000). Z. Naturforsch. Teil B, 55, 995–999.

Wehmschulte, R. J. & Power, P. P. (1997). J. Am. Chem. Soc. 119, 2847-2852.